Synthesis of 1-Thio-Substituted Isoquinoline Derivatives by Tandem Cyclization of Isothiocyanates

Li-Rong Wen,^{†,‡} Qian Dou,[†] Yuan-Chao Wang,[†] Jin-Wei Zhang,[‡] Wei-Si Guo,^{*,†,‡} and Ming Li^{*,†}®

[†]State Key Laboratory Base of Eco-Chemical Engineering, College of Chemistry and Molecular Engineering, Qingdao University of Science and Technology, Qingdao 266042, P. R. China

[‡]State Key Laboratory for Marine Corrosion and Protection, Luoyang Ship Material Research Institute (LSMRI), Qingdao 266101, P. R. China





ABSTRACT: A copper-catalyzed tandem arylation-cyclization process to access 1-(arylthio)isoquinolines from isothiocyanates and diaryliodonium salts is described. It is the first general method to construct the potentially useful 1-(arylthio)isoquinoline derivatives. Moreover, 1-(methylthio)isoquinoline derivatives were also achieved successfully with MeOTf instead of diaryliodonium salts under metal-free conditions. Mechanistic studies reveal that these two processes proceed in different routes. This method has been successfully applied to the synthesis of quinazolinone alkaloid rutaecarpine.

■ INTRODUCTION

Isoquinoline is an important heteroaromatic scaffold existing in natural products, pharmaceuticals, and functional materials,¹ such as berberine alkaloids,² antispasmodic drug papaverine,³ and cytotoxic alkaloid quinocarcin (Figure 1).⁴ Thus, diverse



Figure 1. Representative bioactive isoquinoline natural products.

synthetic methods have been developed for constructing isoquinoline derivatives. However, classical methods of isoquinoline synthesis, involving Bischler–Napieralski⁵ and Pictet–Spengler reactions,⁶ require harsh acidic conditions which limited their usage. Recently, many efficient methods have been reported to construct isoquinolines through annulation reactions catalyzed by transition metals,⁷ such as palladium,⁸ nickel,⁹ silver,¹⁰ copper,¹¹ rhodium,¹² and ruthenium.¹³ Although significant advances have been made toward

isoquinoline construction, to the best of our knowledge, there has been no general method to synthesize 1-(arylthio)isoquinoline derivatives. Given the increasing prevalence of sulfur-containing compounds in pharmaceuticals, ¹⁴ introducing an arylthio or methylthio group into the isoquinoline motif would be of great synthetic value.

Recently, the arylation-cyclization process based on diaryliodonium salts has become a powerful strategy for the construction of heterocycles.¹⁵ Various substrates, such as alkenes,¹⁶ alkynes,¹⁷ and nitriles,¹⁸ can be easily arylated with diaryliodonium salts to generate carbocation intermediates, which undergo cyclization to obtain structure-divergent molecules. Isothiocyanates, as versatile synthetic intermediates, have been widely used in the construction of heterocyclic compounds.¹⁹ Traditionally, reactions occurred at the C atom of isothiocyanates; i.e., the -NCS group was applied as an electrophile. Recently, tandem radical cyclization reactions involving isothiocyanates have received more attention.²⁰ However, reactions induced by the S atom of isothiocyanate have been much less reported.²¹ Moreover, β -arylethyl isothiocyanates and vinyl isothiocyanates, as potentially synthetic useful intermediates, are seldom applied for the synthesis of heterocycles.²² As our continuing interests in using isothiocyanates as the starting materials for the construction of heterocycles,²³ herein, we report a new efficient tandem reaction to synthesize 1-(arylthio)- and 1-(methylthio)-

Received:October 28, 2016Published:January 16, 2017

isoquinoline derivatives through isothiocyanates with diaryliodonium salts and methyl triflate, respectively.

RESULTS AND DISCUSSION

Considering many natural isoquinoline alkaloids containing 3,4dimethoxy or 3,4-methylenedioxy groups (Figure 1), 4-(2isothiocyanatoethyl)-1,2-dimethoxybenzene 1a was chosen as a model substrate to react with diphenyliodonium salt 2a (Table 1). Initially, our previous optimized conditions^{23a} were used,

Table 1. Optimization of the Reaction Conditions^a

| | | MeO、 🔨 🦳 | | | |
|-----------------|------------------|-----------------|-----------------|-----------------|------------------------|
| MeO | Pł | n—ļ—X | catalyst | | ΥΥ) |
| | 🥖 א⊂s ⁺ | Ρ́h | solvent, T (°C) | MeO | |
| Mee | | | | | Ś. Ph |
| 1a | | 2a | | | 3a |
| entry | catalyst (mol %) | Х | solvent | $T [^{\circ}C]$ | yield [%] ^b |
| 1 | CuCl (10) | OTf | DCE | 80 | 34 |
| 2 | $Cu(OTf)_2$ (10) | OTf | DCE | 80 | 28 |
| 3 | $Cu(OAc)_2$ (10) | OTf | DCE | 80 | 12 |
| 4 | CuBr (10) | OTf | DCE | 80 | 26 |
| 5 | CuTC (10) | OTf | DCE | 80 | 28 |
| 6 | CuI (10) | OTf | DCE | 80 | 32 |
| 7 | none | OTf | DCE | 80 | 0 |
| 8 | CuCl (10) | PF ₆ | DCE | 80 | 90 |
| 9 | CuCl (10) | BF_4 | DCE | 80 | 18 |
| 10 | CuCl (10) | PF_6 | THF | 80 | 50 |
| 11 | CuCl (10) | PF_6 | toluene | 80 | 83 |
| 12 | CuCl (10) | PF_6 | DCE | 100 | 77 |
| 13 | CuCl (10) | PF_6 | DCE | 60 | 75 |
| 14 | CuCl (5) | PF_6 | DCE | 80 | 39 |
| 15 ^c | CuCl (10) | PF_6 | DCE | 80 | 78 |

^aReaction conditions: **1a** (0.2 mmol), **2a** (0.3 mmol), solvent (1 mL), 2 h, N₂. ^bIsolated yield based on **3a**. ^cUnder an air atmosphere.

Table 2. Synthesis of Isoquinoline Derivatives $3^{a,b}$

and the desired product 1-(phenylthio)-3,4-dihydroisoquinoline 3a was obtained in 34% yield (entry 1). This result motivated us to test different copper catalysts. However, other copper salts such as Cu(OTf)₂, Cu(OAc)₂, CuBr, CuTC, and CuI did not improve the yield of 3a (entries 2-6). In addition, compound 3a was not observed in the absence of copper salt (entry 7). When hexafluorophosphate (PF_6^-) was used as an anion instead of OTf-, 3a was obtained in excellent yield (entry 8), whereas lower yield was provided with tetrafluoroborate (BF_4^{-}) (entry 9), which showed that the anions of diaryliodonium salts had dramatic influence on the reaction. Additionally, screening some other solvents, such as THF and toluene, revealed that DCE was the best choice (entries 10 and 11). The yield of 3a decreased with elevating the temperature to 100 °C or decreasing to 60 °C (entries 12 and 13). When 5 mol % CuCl was used as catalyst, the yield of 3a dropped to 39% (entry 14). A slightly lower yield was observed when the reaction was conducted under the air atmosphere. Finally, the optimal conditions were established as follows: 1a and 2a (X = PF_6) in a ratio of 1:1.5, CuCl (10 mol %), DCE (0.2 M) at 80 °C for 2 h.

Under the optimized reaction conditions, the scope of substrates was investigated (Table 2). The 1-(arylthio)-3,4-dihydroisoquinolines 3a-f were generated in good to excellent yields (72–96%) from isothiocyanate 1a and diaryliodonium salts 2, except the *para*-OMe substrate 2c (64% yield). The desired products 3g-l were also formed in moderated to good yields when 3,4-methylenedioxy-substituted isothiocyanate 1b was used instead of 1a. In addition, disubstituted diaryliodonium salts, such as 2,4-dimethyl-substituted compound 2m, were also compatible to the reaction, and product 3m was obtained in 82% yield. Disappointingly, no cyclization products were formed when monosubstituted substrates 1 (4-Me, 4-OMe, 4-F-phenyl) were used. To further extend the scope of



"Reaction conditions: 1 (0.4 mmol), 2 (0.6 mmol), CuCl (0.04 mmol), DCE (2 mL), 2 h, N₂. "Isolated yield

Article

Table 3. Synthesis of Isoquinoline Derivatives $4^{a,b}$



^aReaction conditions: 1 (0.4 mmol), MeOTf (0.6 mmol), DCE (2 mL), 2 h, N₂. ^bIsolated yield.





Scheme 2. Control Experiments



Table 4. Synthesis of Monosubstituted Derivatives $3^{a,b}$



^aReaction conditions: 1 (0.4 mmol), 2 (0.6 mmol), CuCl (0.04 mmol), TfOH (0.002 mmol), DCE (2 mL), 2 h, N₂. ^bIsolated yield.

Scheme 3. Proposed Reaction Mechanism



the protocol, several vinyl isothiocyanates were synthesized and reacted with diaryliodonium salts **2**. The reaction proceeded smoothly, and isoquinolines 3n-q were provided in moderate yields. The reason may be subject to the electron-withdrawing COOEt group, which makes the carbon atom of the NCS group more electron-deficient to facilitate the electrophilic cyclization. The structures of **3** were unambiguously confirmed by X-ray crystallographic analysis of compound **3a** (see Figure S1).

Considering that aryl carbocations generated from 2 could react with isothiocyanates 1, we envisioned that methyl carbocation may proceed via a similar process to form 1-(methylthio)isoquinoline derivatives. To test the idea, methyl triflate (MeOTf) was chosen to react with isothiocyanates 1 under the above conditions without copper catalyst (Table 3). To our delight, 1-(methylthio)-3,4-dihydroisoquinolines 4a–f were obtained in 51–86% yields. The reactions showed obvious electron effects, and higher yield was obtained with a highly electron-rich substrate (4f, 86%), whereas the electron-deficient 4-CF₃ substrate provided lower yield (4e, 51%). Notably, heterocyclic thiophene and indole substrates were also suitable for the reaction; the desired products 4g and 4h were obtained in 60% and 84% yields, respectively.

Interestingly, when vinyl isothiocyanate **1n** was used to react with MeOTf, 1-(methylthio)isoquinoline **4i** was not obtained, whereas 1-(phenylthio)isoquinoline **3n** was provided in 46% yield (Scheme 1). Furthermore, **4b** was produced in 72% yield when isothiocyanate **1r** was used, whereas **3r** was not formed. These phenomena suggested that the two reactions maybe proceed through different pathways.

In order to explain these phenomena, a series of control experiments were conducted (Scheme 2). Isothiocyanate 1r

could be cyclized smoothly to provide thioamide 5 in the presence of trifluoromethanesulfonic acid (TfOH) (eq 1). Then, the reaction of 5 with 2a was performed under the standard conditions, and the dihydroisoquinoline 3r was obtained in 79% yield (eq 2). However, when the reaction of 1r with 2a was directly performed under the standard conditions, no cyclized product 3r was obtained, but thiocarbamate 6 was formed in 68% yield (eq 3). On the basis of the above results, we performed eqs 1 and 2 in one-pot in the presence of stoichiometric TfOH (eq 4), and 3r was obtained, albeit in a low yield of 24%. Considering a trace amount of TfOH is sufficient to initiate the intramolecular electrophilic cyclization, 0.5 mol % TfOH was added in the reaction mixture; to our delight, the yield of 3r increased to 75%. Thus, a cyclization/arylation process was proposed for the synthesis of monosubstituted product 3r.

Otherwise, **4b** was generated in 80% yield from **5** under the standard conditions (eq 5). When vinyl isothiocyanate **1n** was used under the same conditions as eq 1, the thioamide was not obtained and only starting material **1n** was recovered (eq 6). These results indicated that the cyclization would occur prior to the methylation for the reactions of isothiocyanates with MeOTf, which was probably due to a trace amount of TfOH involving in MeOTf (MeOTf easily hydrolyzes to form TfOH). To verify this hypothesis, the reaction of **1r** with MeOTf was conducted with 10 mol % 2,6-di-*tert*-butylpyridine as a base, and it was completely prevented. These results confirmed that TfOH should play an important role in the cyclization process.

To further examine the generality of the reaction in Scheme 2, eq 4, other monosubstituted substrates (4-OMe, 4-F, 3-OMe) were investigated (Table 4); the results showed that the

reactions could proceeded smoothly and the cyclized monosubstituted products **3s-u** were formed in good yields.

On the basis of the above results, a possible mechanism is proposed in Scheme 3. In path 1, when Ar_2IPF_6 reacts with isothiocyanate 1 in the absence of TfOH, the sulfonium intermediate A is formed initially through S-arylation of the generated aryl carbocation.^{23a} Subsequently, the intermediate A undergoes intramolecular Friedel–Crafts-type cyclization, followed by deprotonation, to provide the desired 1-(arylthio)isoquinolines 3a-q (path 1). In path 2, when MeOTf reacts with 1, the NCS group is initially protonated, followed by cyclization, to generate intermediate D. Finally, a methylation of D with MeOTf gives products 4. The reaction mechanism for the synthesis of 3r-u is similar to that of compounds 4.

The quinazolinone alkaloid rutaecarpine, isolated in 1915 from *Evodia rutaecarpa*, was used in Traditional Chinese Medicine for the treatment of inflammation-related disorders.²⁴ To further illustrate the potential utility of this new method, a concise total synthesis of rutaecarpine was explored (Scheme 4). Initially, isothiocyanate **8** was synthesized from commer-

Scheme 4. Application in the Synthesis of Rutaecarpine



cially available tryptamine 7. Under the above standard cyclization conditions, the methylthio-substituted 3,4-dihydro- β -carboline 9 was generated in 90% yield. Finally, annulation with 2-aminobenzoic acid afforded the natural alkaloid rutaecarpine.^{24b} The facile synthetic route could provide a pathway for rutaecarpine analogues synthesis.

In summary, we have developed a new tandem reaction to synthesize 1-thio-substituted isoquinolines from isothiocyanates with diaryliodonium salts or methyl triflate under mild conditions. Control experiments suggest that two mechanistically different pathways might be involved. To the best of our knowledge, this is the first general method to construct 1-(arylthio)isoquinolines. In addition, the method was applied as a key step to the short synthesis of alkaloid rutaecarpine. Further studies to explore new synthetic methods using isothiocyanates are in progress.

EXPERIMENTAL SECTION

General Information. All air- or moisture-sensitive reactions were conducted under a nitrogen atmosphere. DCE was distilled from CaH₂. Unless noted, all commercial reagents were used without further purification. Melting points were recorded on a microscopic melting apparatus and uncorrected. ¹H NMR spectra were recorded at 500 MHz, and ¹³C NMR spectra were recorded at 125 MHz in CDCl₃ or DMSO-*d*₆. Chemical shifts were reported in parts per million (δ) relative to tetramethylsilane (TMS). HRMS was obtained on a spectrometer with an ESI source. The X-ray single-crystal diffraction was performed on a CCD area detector. Silica gel (200–300 mesh) was used for column chromatography and silica GF254 for TLC. **Preparation of Starting Materials.** Diaryliodonium salts²⁵ were prepared according to previously published procedures. The (2-isothiocyanatoethyl) benzenes²⁶ and ethyl (*Z*)-2-isothiocyanato-3-phenylacrylates^{22a} were also synthesized according to previously published procedures.

General Procedure for the Synthesis of Compounds 3 (3a for Example). Under a N_2 atmosphere, a dry Schlenk tube was charged with a mixture of 4-(2-isothiocyanatoethyl)-1,2-dimethoxybenzene 1a (90 mg, 0.40 mmol), diphenyliodonium salt 2a (256 mg, 0.60 mmol), CuCl (4.0 mg, 0.04 mmol), and DCE (2.0 mL). The mixture was allowed to stir at 80 °C for 2 h. After completion, the mixture was cooled to room temperature, quenched with saturated NaHCO₃, and extracted with EtOAc (10 mL × 3). The organic layer was washed with saturated NaCl and dried over anhydrous MgSO₄. Evaporation of the solvent, followed by purification on silica gel (petroleum ether/ethyl acetate), provided product 3a.

6,7-Dimethoxy-1-(phenylthio)-3,4-dihydroisoquinoline (**3a**). Isolated yield 108 mg (90%); white solid; mp 86–88 °C; R_f 0.20 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.68 (t, *J* = 7.4 Hz, 2H), 3.66 (t, *J* = 7.4 Hz, 2H), 3.92 (s, 3H), 3.94 (s, 3H), 6.72 (s, 1H), 7.31 (s, 1H), 7.36–7.42 (m, 3H), 7.57 (d, *J* = 7.1 Hz, 2H); ¹³C NMR (CDCl₃, 125 MHz): δ 25.9, 48.6, 55.9, 56.1, 108.6, 110.1, 121.5, 128.5, 128.9, 129.8, 131.1, 134.8, 147.3, 151.1, 163.3; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₇H₁₇NO₂S, 300.1053, found 300.1054

1-((4-Fluorophenyl)thio)-6,7-dimethoxy-3,4-dihydroisoquinoline (**3b**). Isolated yield 122 mg (96%); white solid; mp 105–106 °C; R_f 0.24 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.65 (t, *J* = 7.4 Hz, 2H), 3.63 (t, *J* = 7.2 Hz, 2H), 3.92 (s, 3H), 3.93 (s, 3H), 6.70 (s, 1H), 7.09 (t, *J* = 8.6 Hz, 2H), 7.28 (s, 1H), 7.51–7.54 (m, 2H); ¹³C NMR (CDCl₃, 125 MHz): δ 26.0, 48.6, 56.0, 56.1, 108.3, 110.1, 116.2 (d, ²*J*_{C-F} = 21.9 Hz), 121.5, 124.6, 131.1, 137.2 (d, ¹*J*_{C-F} = 248.3 Hz), 147.4, 151.2, 162.1, 163.1, 164.0; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₇H₁₆FNO₂S, 318.0959, found 318.0956.

6,7-Dimethoxy-1-((4-methoxyphenyl)thio)-3,4-dihydroisoquinoline (3c). Isolated yield 85 mg (64%); white solid; mp 138–140 °C; R_f 0.10 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.64 (t, *J* = 7.0 Hz, 2H), 3.62 (t, *J* = 7.0 Hz, 2H), 3.82 (s, 3H), 3.92 (s, 3H), 3.93 (s, 3H), 6.69 (s, 1H), 6.93 (d, *J* = 7.9 Hz, 2H), 7.31 (s, 1H), 7.47 (d, *J* = 7.9 Hz, 2H); ¹³C NMR (CDCl₃, 125 MHz): δ 26.1, 48.63, 55.27, 56.0, 56.2, 108.6, 110.2, 114.8, 120.0, 121.8, 131.1, 136.8, 147.5, 151.2, 160.2; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₈H₁₉NO₃S, 330.1164, found 330.1152.

6,7-Dimethoxy-1-(p-tolylthio)-3,4-dihydroisoquinoline (**3d**). Isolated yield 110 mg (88%); white solid; mp 129–131 °C; R_f 0.22 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.36 (s, 3H), 2.65 (t, *J* = 7.4 Hz, 2H), 3.64 (t, *J* = 7.2 Hz, 2H), 3.91 (s, 3H), 3.92 (s, 3H), 6.69 (s, 1H), 7.21 (d, *J* = 7.9 Hz, 2H), 7.30 (s, 1H), 7.43 (d, *J* = 8.0 Hz, 2H); ¹³C NMR (CDCl₃, 125 MHz): δ 21.3, 26.0, 48.6, 56.0, 56.1, 108.4, 110.0, 121.6, 125.9, 130.0, 131.1, 135.0, 138.8, 147.3, 151.0, 163.5; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₈H₁₉NO₂S, 314.1210, found 314.1212.

6,7-Dimethoxy-1-(o-tolylthio)-3,4-dihydroisoquinoline (**3e**). Isolated yield 101 mg (81%); white solid; mp 94–96 °C; R_f 0.23 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.43 (s, 3H), 2.65 (t, *J* = 7.4 Hz, 2H), 3.62 (t, *J* = 7.2 Hz, 2H), 3.92 (s, 6H), 6.70 (s, 1H), 7.20–7.23 (m, 1H), 7.30–7.32 (m, 3H), 7.55 (d, *J* = 7.6 Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ 20.9, 26.0, 48.6, 55.9, 56.1, 108.6, 110.1, 121.7, 126.5, 128.9, 129.4, 130.6, 131.1, 136.2, 142.6, 147.4, 151.0, 162.5; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₈H₁₉NO₂S, 314.1210, found 314.1211.

6,7-Dimethoxy-1-(m-tolylthio)-3,4-dihydroisoquinoline (**3f**). Isolated yield 90 mg (72%); white solid; mp 93–95 °C; R_f 0.24 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.36 (s, 3H), 2.66 (t, *J* = 7.4 Hz, 2H), 3.65 (t, *J* = 7.2 Hz, 2H), 3.90 (s, 3H), 3.92 (s, 3H), 6.70 (s, 1H), 7.16 (d, *J* = 7.5 Hz, 1H), 7.26–7.30 (m, 2H), 7.35–7.37 (m, 2H); ¹³C NMR (CDCl₃, 125 MHz): δ 21.3, 26.0, 48.6, 56.0, 56.1, 108.6, 110.1, 121.6, 128.8, 129.3, 129.5, 131.1, 132.0, 135.4, 138.7, 147.4, 151.1, 163.4; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₈H₁₉NO₂S, 314.1210, found 314.1214.

5-(Phenylthio)-7,8-dihydro-[1,3]dioxolo[4,5-g]isoquinoline (**3g**). Isolated yield 59 mg (52%); white solid; mp 154–156 °C; R_f 0.39 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.63 (t, *J* = 7.3 Hz, 2H), 3.60 (t, *J* = 7.3 Hz, 2H), 6.00 (s, 2H), 6.68 (s, 1H), 7.31 (s, 1H), 7.34–7.41 (m, 3H), 7.55 (d, *J* = 6.9 Hz, 2H); ¹³C NMR (CDCl₃, 125 MHz): δ 26.7, 48.4, 80.7, 84.1, 101.4, 105.9, 107.8, 122.9, 128.6, 129.0, 133.0, 135.0, 146.4, 149.5, 163.2; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₆H₁₃NO₂S, 284.0745, found 284.0752.

5-(*p*-Tolylthio)-7,8-dihydro-[1,3]dioxolo[4,5-g]isoquinoline (**3h**). Isolated yield 60 mg (50%); white solid; mp 161–163 °C; R_f 0.40 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.36 (s, 3H), 2.61 (t, *J* = 7.3 Hz, 2H), 3.59 (t, *J* = 7.3 Hz, 2H), 5.99 (s, 2H), 6.67 (s, 1H), 7.20 (d, *J* = 7.8 Hz, 2H), 7.31 (s, 1H), 7.43 (d, *J* = 7.9 Hz, 2H); ¹³C NMR (CDCl₃, 125 MHz): δ 26.7, 48.5, 101.4, 105.9, 107.8, 130.0, 125.8, 129.9, 130.0, 135.2, 138.8, 146.3, 149.3, 163.3; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₇H₁₅NO₂S, 298.0902, found 298.0912.

5-((4-Methoxyphenyl)thio)-7,8-dihydro-[1,3]dioxolo[4,5-g]isoquinoline (**3**i). Isolated yield 91 mg (73%); white solid; mp 153– 155 °C; R_f 0.28 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.61 (t, *J* = 7.3 Hz, 2H), 3.58 (t, *J* = 7.3 Hz, 2H), 3.82 (s, 3H), 5.99 (s, 2H), 6.67 (s, 1H), 6.94 (d, *J* = 8.9 Hz, 2H), 7.31 (s, 1H), 7.46 (d, *J* = 8.5 Hz, 2H); ¹³C NMR (CDCl₃, 125 MHz): δ 26.8, 48.5, 55.3, 101.4, 105.8, 107.8, 114.8, 119.8, 123.0, 133.0, 137.0, 146.4, 149.3, 160.2, 163.6; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₇H₁₅NO₃S, 314.0851, found 314.0862.

5-((4-Fluorophenyl)thio)-7,8-dihydro-[1,3]dioxolo[4,5-g]isoquinoline (**3***j*). Isolated yield 82 mg (68%); white solid; mp 156–158 °C; R_f 0.62 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.62 (t, *J* = 7.3 Hz, 2H), 3.58 (t, *J* = 7.3 Hz, 2H), 6.00 (s, 2H), 6.68 (s, 1H), 7.09 (t, *J* = 8.9 Hz, 2H), 7.28 (s, 1H), 7.50–7.52 (m, 2H); ¹³C NMR (CDCl₃, 125 MHz): δ 26.7, 48.5, 101.4, 105.7, 107.9, 116.2 (d, ²*J*_{C-F} = 21.9 Hz), 122.8, 124.6, 133.0, 137.2, 137.3, 146.4, 149.5, 162.9, 163.1 (d, ¹*J*_{C-F} = 248.3 Hz); HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₆H₁₂FNO₂S, 302.0651, found 302.0659.

5-((2-Fluorophenyl)thio)-7,8-dihydro-[1,3]dioxolo[4,5-g]isoquinoline (**3k**). Isolated yield 89 mg (74%); white solid; mp 159–161 °C; R_f 0.52 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.64 (t, *J* = 7.3 Hz, 2H), 3.61 (t, *J* = 7.3 Hz, 2H), 6.02 (s, 2H), 6.70 (s, 1H), 7.15–7.22 (m, 2H), 7.40–7.44 (m, 1H), 7.58 (t, *J* = 6.7 Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ 26.6, 48.5, 101.4, 105.9, 107.8, 116.0 (d, ²*J*_{C-F} = 22.7 Hz), 117.0 (d, ²*J*_{C-F} = 18.1 Hz), 122.7, 124.5, 131.3, 133.0, 137.2, 146.4, 149.5, 161.3, 162.7 (d, ¹*J*_{C-F} = 253.0 Hz); HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₆H₁₂FNO₂S, 302.0651, found 302.0661.

5-((3-Fluorophenyl)thio)-7,8-dihydro-[1,3]dioxolo[4,5-g]isoquinoline (**3**). Isolated yield 72 mg (60%); white solid; mp 125–127 °C; R_f 0.58 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.64 (t, *J* = 7.4 Hz, 2H), 3.62 (t, *J* = 7.4 Hz, 2H), 6.00 (s, 2H), 6.68 (s, 1H), 7.05 (t, *J* = 8.1 Hz, 1H), 7.28–7.35 (m, 4H); ¹³C NMR (CDCl₃, 125 MHz): δ 26.6, 48.5, 101.4, 105.9, 107.9, 115.6 (d, ${}^{2}J_{CF}$ = 21.1 Hz), 121.6 (d, ${}^{2}J_{CF}$ = 22.9 Hz), 122.8, 130.0, 131.8, 133.1, 146.5, 149.6, 162.4, 162.5 (d, ${}^{1}J_{CF}$ = 247.3 Hz); HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₆H₁₂FNO₂S, 302.0651, found 302.0648.

5-((2,4-Dimethylphenyl)thio)-7,8-dihydro-[1,3]dioxolo[4,5-g]isoquinoline (**3m**). Isolated yield 102 mg (82%); white solid; mp 118–120 °C; R_f 0.40 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.36 (s, 3H), 2.40 (s, 3H), 2.62 (t, *J* = 7.3 Hz, 2H), 3.59 (t, *J* = 7.3 Hz, 2H), 6.02 (s, 2H), 6.70 (s, 1H), 7.06 (d, *J* = 8.0 Hz, 1H), 7.15 (s, 1H), 7.38 (s, 1H), 7.45 (d, *J* = 7.8 Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ 20.8, 21.3, 48.5, 101.3, 105.9, 107.8, 123.0, 125.1, 127.4, 131.5, 132.9, 136.4, 139.6, 142.6, 146.3, 149.3, 162.5; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₈H₁₇NO₂S, 312.1058, found 312.1062.

Ethyl 1-(Phenylthio)isoquinoline-3-carboxylate (3n). Isolated yield 63 mg (51%); white solid; mp 63–65 °C; R_f 0.68 (petroleum ether/ethyl acetate = 10:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 1.33 (t, *J* = 7.2 Hz, 3H), 2.40 (q, *J* = 7.1 Hz, 2H), 7.38–7.44 (m, 3H), 7.69–7.78 (m, 4H), 7.93 (d, *J* = 8.0 Hz, 1H), 8.29 (s, 1H), 8.38 (d, *J* =

8.2 Hz, 1H); ^{13}C NMR (CDCl₃, 125 MHz): δ 14.1, 29.6, 61.3, 121.5, 124.9, 128.1, 128.5, 128.8, 129.4, 129.9, 131.0, 134.6, 135.5, 141.0, 159.6, 165.5; HRMS (ESI-TOF, $[M + H]^+$): Calcd for $C_{18}H_{15}NO_2S$, 310.0902, found 310.0914.

Ethyl 7-*Methyl*-1-(*phenylthio*)*isoquinoline*-3-*carboxylate* (**3***o*). Isolated yield 75 mg (58%); white solid; mp 84–86 °C; R_f 0.56 (petroleum ether/ethyl acetate = 10:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 1.32 (t, *J* = 7.3 Hz, 3H), 2.60 (s, 3H), 4.32 (q, *J* = 7.2 Hz, 2H), 7.38–7.44 (m, 3H), 7.59 (d, *J* = 8.5 Hz, 1H), 7.71 (d, *J* = 7.9 Hz, 2H), 7.82 (d, *J* = 7.9 Hz, 1H), 8.14 (s, 1H), 8.26 (s, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ 14.1, 22.2, 61.2, 121.4, 124.0, 127.2, 128.4, 128.5, 128.8, 130.0, 133.1, 133.6, 134.6, 140.1, 140.3, 158.5, 165.6; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₉H₁₇NO₂S, 324.1058, found 324.1049.

Ethyl 7-*Methoxy*-1-(*phenylthio*)*isoquinoline*-3-*carboxylate* (**3***p*). Isolated yield 82 mg (60%); white solid; mp 112–114 °C; R_f 0.50 (petroleum ether/ethyl acetate = 10:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 1.33 (t, *J* = 7.0 Hz, 3H), 2.60 (s, 3H), 4.33 (q, *J* = 7.1 Hz, 2H), 7.35–7.42 (m, 4H), 7.56 (s, 1H), 7.71 (d, *J* = 8.0 Hz, 2H), 7.83 (d, *J* = 9.2 Hz, 1H), 8.27 (s, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ 14.1, 55.6, 61.2, 121.6, 123.7, 128.2, 128.8, 129.9, 130.3, 130.8, 134.0, 139.4, 157.1, 160.3, 165.6; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₉H₁₇NO₃S, 340.1007, found 340.1012.

Ethyl 1-((4-Methoxyphenyl)thio)-7-methylisoquinoline-3-carboxylate (**3q**). Isolated yield 89 mg (63%); white solid; mp 89–91 °C; R_f 0.55 (petroleum ether/ethyl acetate = 10:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 1.32 (t, *J* = 7.3 Hz, 3H), 2.60 (s, 3H), 3.84 (s, 3H), 4.30 (q, *J* = 7.2 Hz, 2H), 6.99 (d, *J* = 8.5 Hz, 2H), 7.58 (d, *J* = 8.5 Hz, 1H), 7.63 (d, *J* = 8.5 Hz, 2H), 7.80 (d, *J* = 7.9 Hz, 1H), 8.13 (s, 1H), 8.22 (s, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ 14.0, 22.1, 55.3, 61.1, 114.5, 120.1, 121.0, 123.8, 128.0, 128.5, 133.0, 133.5, 136.9, 139.9, 140.3, 159.4, 160.2, 165.7; HRMS (ESI-TOF, [M + H]⁺): calcd for C₂₀H₁₉NO₃S, 354.1158, found, 354.1171.

General Procedure for the Synthesis of Compounds 4 (4a for Example). A dry Schlenk tube was charged with a mixture of (2isothiocyanatoethyl) benzene 1c (66 mg, 0.40 mmol), methyltriflate (99 mg, 0.60 mmol), and DCE (2.0 mL). The mixture was allowed to stir at 80 °C for 2 h under a N₂ atomsphere. After that, the mixture was cooled to room temperature, quenched with saturated NaHCO₃, and extracted with EtOAc (10 mL × 3). The organic layer was washed with saturated NaCl and dried over anhydrous MgSO₄. Evaporation of the solvent, followed by purification on silica gel (petroleum ether/ ethyl acetate), provided the product 4a.

1-(Methylthio)-3,4-dihydroisoquinoline (4a).^{22b} Isolated yield 46 mg (64%); colorless oil; R_f 0.48 (petroleum ether/ethyl acetate = 10:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.44 (s, 3H), 2.74 (t, *J* = 7.3 Hz, 2H), 3.76 (t, *J* = 7.3 Hz, 2H), 7.19 (d, *J* = 7.3 Hz, 1H), 7.28 (t, *J* = 7.3 Hz, 1H), 7.36 (t, *J* = 7.6 Hz, 1H), 7.66 (d, *J* = 7.3 Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ 12.2, 26.5, 48.2, 124.7, 126.9, 127.3, 129.3, 130.8, 136.8, 164.0; HRMS (ESI-TOF, $[M + H]^+$): calcd for C₁₀H₁₁NS, 178.0685, found 178.0689.

7-Methyl-1-(methylthio)-3,4-dihydroisoquinoline (**4b**). Isolated yield 55 mg (72%); colorless oil; R_f 0.43 (petroleum ether/ethyl acetate = 10:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.36 (s, 3H), 2.44 (s, 3H), 2.68 (t, *J* = 7.3 Hz, 2H), 3.74 (t, *J* = 7.3 Hz, 2H), 7.04 (d, *J* = 7.6 Hz, 1H), 7.17 (d, *J* = 7.5 Hz, 1H), 7.47 (s, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ 12.2, 21.2, 26.2, 48.5, 125.3, 127.2, 129.2, 131.5, 133.8, 136.5, 164.1; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₁H₁₄NS, 192.0841, found 192.0837.

7-Methoxy-1-(methylthio)-3,4-dihydroisoquinoline (*4c*). Isolated yield 62 mg (75%); colorless oil; R_f 0.40 (petroleum ether/ethyl acetate = 10:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.44 (s, 3H), 2.66 (t, *J* = 7.3 Hz, 2H), 3.73 (t, *J* = 7.3 Hz, 2H), 3.82 (s, 3H), 6.92 (q, *J* = 3.7 Hz, 1H), 7.11 (d, *J* = 8.5 Hz, 1H), 7.21 (d, *J* = 1.8 Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ 12.4, 25.7, 48.7, 55.5, 110.0, 116.7, 128.2, 129.0, 130, 158.5, 164.0; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₁H₁₃NOS, 208.0791, found 208.0795.

7-Fluoro-1-(methylthio)-3,4-dihydroisoquinoline (4d). Isolated yield 44 mg (56%); white solid; mp 49–51 °C; R_f 0.50 (petroleum ether/ethyl acetate = 10:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.44

(s, 3H), 2.70 (t, J = 7.0 Hz, 2H), 3.75 (t, J = 7.3 Hz, 2H), 7.07 (td, J = 7.9, 2.4 Hz, 1H), 7.16 (t, J = 7.0 Hz, 1H), 7.37 (dd, J = 9.2, 2.4 Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ 12.4, 25.8, 48.4, 111.9 (d, ² $J_{CF} = 23.3$ Hz), 117.6 (d, ² $J_{CF} = 20.9$ Hz), 128.7 (d, ³ $J_{CF} = 7.0$ Hz), 130.4 (d, ³ $J_{CF} = 7.0$ Hz), 132.4 (d, ⁴ $J_{C-F} = 3.0$ Hz), 161.5 (d, ¹ $J_{CF} = 245.3$ Hz), 163.1; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₀H₁₀FNS, 196.0591, found 196.0594.

1-(*Methylthio*)-7-(*trifluoromethyl*)-3,4-*dihydroisoquinoline* (*4e*). Isolated yield 50 mg (51%); white solid; mp 46–48 °C; R_f 0.46 (petroleum ether/ethyl acetate = 10:1 v/v). ¹H NMR (CDCl₃, 500 MHz): 2.45 (s, 3H), 2.79 (t, *J* = 7.0 Hz, 2H), 3.78 (t, *J* = 7.3 Hz, 2H), 7.32 (d, *J* = 7.9 Hz, 1H), 7.61 (d, *J* = 7.9 Hz, 1H), 7.89 (s, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ 12.3, 26.5, 47.8, 121.7, 123.8 (d, ¹*J* = 272.1 Hz), 127.3, 127.9, 129.2 (d, ²*J* = 32.7 Hz), 129.5, 140.7, 162.9; HRMS (ESI-TOF, $[M + H]^+$): calcd for C₁₁H₁₁F₃NS, 246.0559, found 246.0557

6,7-Dimethoxy-1-(methylthio)-3,4-dihydroisoquinoline (4f).^{22b} Isolated yield 82 mg (86%); white solid; mp 92–94 °C; R_f 0.37 (petroleum ether/ethyl acetate = 10:1 v/v). ¹H NMR (CDCl₃, 500 MHz):^{22b} δ 2.43 (s, 3H), 2.65 (t, *J* = 7.3 Hz, 2H), 3.71 (t, *J* = 7.0 Hz, 2H), 3.90 (s, 6H), 6.68 (s, 1H), 7.17 (s, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ 12.3, 26.2, 48.3, 55.9, 56.1, 108.1, 110.0, 122.1, 130.4, 147.5, 151.0, 163.4; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₂H₁₆NO₂S, 238.0896, found 238.0894.

4-(*Methylthio*)-6,7-dihydrothieno[3,2-c]pyridine (**4g**).²⁷ Isolated yield 44 mg (60%); colorless oil; R_f 0.65 (petroleum ether/ethyl acetate = 10:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.44 (s, 3H), 2.86 (t, *J* = 7.9 Hz, 2H), 3.85 (t, *J* = 7.9 Hz, 2H), 7.07 (d, *J* = 5.5 Hz, 1H), 7.15 (d, *J* = 5.5 Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ 11.8, 22.8, 49.0, 122.2, 123.3, 131.1, 141.9, 160.2; HRMS (ESI-TOF, [M + H]⁺): calcd for C₈H₉NS₂, 184.0249, found 184.0252.

9-Methyl-1-(methylthio)-4,9-dihydro-3H-pyrido[3,4-b]indole (**4h**). Isolated yield 78 mg (84%); white solid; mp 55–58 °C; R_f 0.60 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.54 (s, 3H), 2.85 (t, *J* = 7.6 Hz, 2H), 3.87 (t, *J* = 7.6 Hz, 2H), 4.04 (s, 3H), 7.14–7.18 (m, 1H), 7.34 (d, *J* = 3.7 Hz, 2H), 7.61 (d, *J* = 7.9 Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ 12.2, 20.0, 32.5, 49.6, 110.2, 117.4, 119.9, 120.0, 124.2, 124.3, 130.0, 138.5, 156.4; HRMS (ESI-TOF, $[M + H]^+$): calcd for C₁₃H₁₄N₂S, 231.0950, found 231.0956.

Procedure for the Synthesis of Compound 5. Under a N₂ atmosphere, a dry Schlenk tube was charged with a mixture of 1-(2-isothiocyanatoethyl)-4-methylbenzene 1r (71 mg, 0.40 mmol), TfOH (90 mg, 0.60 mmol), and DCE (2.0 mL). The mixture was allowed to stir at 80 °C for 2 h. After completion, the mixture was cooled to room temperature, neutralized with saturated NaHCO₃, and extracted with EtOAc (10 mL × 3). The organic layer was washed with saturated NaCl and dried over anhydrous MgSO₄. Evaporation of the solvent, followed by purification on silica gel (petroleum ether/ethyl acetate), provided the product 5 as an oily yellow solid.

7-Methyl-3,4-dihydroisoquinoline-1(2H)-thione (5).²⁸ Isolated yield 45 mg (64%); oily yellow solid; R_f 0.40 (petroleum ether/ ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.38 (s, 3H), 2.96 (t, *J* = 6.8 Hz, 2H), 3.50–3.54 (m, 2H), 7.05 (d, *J* = 7.9 Hz, 1H), 7.24 (s, 1H), 8.33 (s, 1H), 8.43 (br s, 1H).

General Procedure for the Synthesis of Compounds 3r-u (3r for Example). Under a N₂ atmosphere, a dry Schlenk tube was charged with a mixture of 1r (71 mg, 0.40 mmol), diphenyliodonium salt 2a (256 mg, 0.60 mmol), CuCl (4 mg, 0.04 mmol), TfOH (0.18 μ L, 0.002 mmol), and DCE (2.0 mL). The mixture was allowed to stir at 80 °C for 2 h. After completion, the mixture was cooled to room temperature, quenched with saturated NaHCO₃, and extracted with EtOAc (10 mL × 3). The organic layer was washed with saturated NaCl and dried over anhydrous MgSO₄. Evaporation of the solvent, followed by purification on silica gel (petroleum ether/ethyl acetate), provided the product 3r as a colorless oil (76 mg, 75%).

7-Methyl-1-(phenylthio)-3,4-dihydroisoquinoline (*3r*). Isolated yield 76 mg (75%); colorless oil, R_f 0.62 (petroleum ether/ethyl acetate = 10:1 v/v). ¹H NMR (DMSO- d_{6i} 500 MHz): δ 2.37 (s, 3H), 2.62 (t, J = 7.2 Hz, 2H), 3.50 (t, J = 7.3 Hz, 2H), 7.20 (d, J = 7.9 Hz,

1H), 7.27 (d, J = 7.6 Hz, 1H), 7.41 (m, 3H), 7.51 (m, 3H); ¹³C NMR (DMSO- d_6 , 125 MHz): δ 21.2, 25.7, 48.4, 115.7, 122.8, 125.0, 125.1, 127.8, 128.1, 128.5, 129.2, 129.4, 129.6, 129.8, 132.3, 132.5, 134.6, 135.7, 136.8, 162.5; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₆H₁₆NS, 254.0998, found 254.0997.

7-Methoxy-1-(phenylthio)-3,4-dihydroisoquinoline (**3***s*). Isolated yield 83 mg (77%); white solid; mp 75–77 °C; R_{*j*} 0.47 (petroleum ether/ethyl acetate = 10:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.65 (t, *J* = 7.3 Hz, 2H), 3.65 (t, *J* = 7.0 Hz, 2H), 3.84 (s, 3H), 6.94 (d, *J* = 7.9 Hz, 1H), 7.12 (d, *J* = 7.9 Hz, 1H), 7.34 (s, 1H), 7.39 (q, *J* = 8.1 Hz, 3H), 7.56 (d, *J* = 6.7 Hz, 2H). ¹³C NMR (CDCl₃, 125 MHz): δ 2.5.5, 48.9, 55.5, 110.3, 117.0, 128.3, 128.6, 129.0, 129.5, 135.1, 158.4, 163.6. HRMS (ESI-TOF, $[M + H]^+$): calcd for C₁₆H₁₆NOS, 270.0947, found 270.0947.

7-*Fluoro-1-(phenylthio)-3,4-dihydroisoquinoline* (**3t**). Isolated yield 74 mg (72%); white solid; mp 89–91 °C; R_f 0.58 (petroleum ether/ethyl acetate = 10:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.70 (t, *J* = 7.0 Hz, 2H), 3.68 (t, *J* = 7.3 Hz, 2H), 7.11 (td, *J* = 8.3, 2.3 Hz, 1H), 7.20 (t, *J* = 6.7 Hz, 1H), 7.37–7.45 (m, 3H), 7.53 (dd, *J* = 9.0, 2.2 Hz, 1H), 7.58 (d, *J* = 6.5 Hz, 2H). ¹³C NMR (CDCl₃, 125 MHz): δ 25.6, 48.7, 112.2 (d, ²*J* = 23.3 Hz), 117.8 (d, ²*J* = 21.2 Hz), 128.9, 129.1, 130.0, 133.0, 134.7, 135.2, 161.5 (d, ¹*J* = 242.7 Hz), 162.8. HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₅H₁₃FNS, 258.0747, found 258.0746.

6-Methoxy-1-(phenylthio)-3,4-dihydroisoquinoline (**3u**). Isolated yield 86 mg (80%); white solid; mp 102–104 °C; R_f 0.53 (petroleum ether/ethyl acetate = 10:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.72 (t, *J* = 7.3 Hz, 2H), 3.66 (t, *J* = 7.3 Hz, 2H), 3.86 (s, 3H), 6.74 (s, 1H), 6.82 (dd, *J* = 8.5, 2.2 Hz, 1H), 7.36–7.42 (m, 3H), 7.58 (d, *J* = 7.3 Hz, 2H), 7.76 (d, *J* = 8.5 Hz, 1H). ¹³C NMR (CDCl₃, 125 MHz): δ 26.9, 48.4, 55.4, 111.8, 111.9, 122.4, 127.0, 128.9, 135.0, 139.7, 161.5, 163.3. HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₆H₁₆NOS, 270.0947, found 270.0948.

Procedure for the Synthesis of Compound 6. Under a N₂ atmosphere, a dry Schlenk tube was charged with a mixture of 1-(2-isothiocyanatoethyl)-4-methylbenzene Ir (71 mg, 0.40 mmol), diphenyliodonium salt 2a (256 mg, 0.60 mmol), CuCl (4 mg, 0.04 mmol), and DCE (2.0 mL). The mixture was allowed to stir at 80 °C for 2 h. After completion, the mixture was cooled to room temperature, quenched with saturated NaHCO₃, and extracted with EtOAc (10 mL × 3). The organic layer was washed with saturated NaCl and dried over anhydrous MgSO₄. Evaporation of the solvent, followed by purification on silica gel (petroleum ether/ethyl acetate), provided the product 6 as a white solid (62 mg, 68%).

S-Phenyl 4-Methylphenethylcarbamothioate (6). Isolated yield 62 mg (68%); White solid; mp 71–73 °C; R_f 0.45 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.33 (s, 3H), 2.74 (s, 2H), 3.49 (d, *J* = 6.9 Hz, 2H), 5.29 (brs, 1H), 6.98 (d, *J* = 6.4 Hz, 2H), 7.08 (d, *J* = 6.8 Hz, 2H), 7.38 (m, 3H), 7.49 (d, *J* = 6.8 Hz, 2H); ¹³C NMR (CDCl₃, 125 MHz): δ 21.0, 35.0, 42.5, 128.5, 129.4 129.5, 129.6, 135.1, 135.5, 136.1, 166.1; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₆H₁₇NOS, 272.1109, found 272.1112.

Procedure for the Synthesis of Rutaecarpine. A round-bottom flask was charged with a mixture of 2-(1H-indol-3-yl) ethan-1-amine 7 (0.80 g, 5.0 mmol), DABCO (4.40 g, 20.0 mmol), and toluene (25 mL). To the stirred mixture was added CS₂ (1.52 g, 20.0 mmol) dropwise, and the mixture was stirred for overnight at room temperature. Then, the dithiocarbamate salt was filtered and washed with toluene. To a round-bottom flask was added the dithiocarbamate salt and CHCl₃ (15 mL). The mixture was cooled to 0 °C, and BTC (0.50 g, dissolved in CHCl₃) was added dropwise. The mixture was stirred for 1 h at 0 °C and refluxed for 2 h at 80 °C. After completion, the mixture was evaporated under reduced pressure and purified by silica gel column chromatography (petroleum ether/ethyl acetate) to obtain the 3-(2-isothiocyanatoethyl)-1H-indole **8**.

3-(2-lsothiocyanatoethyl)-1H-indole (8).²⁹ Isolated yield 858 mg (85%); white solid; mp 45–47 °C (lit. mp 46–47 °C); R_f 0.58 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 3.16 (t, J = 6.8 Hz, 2H), 3.76 (t, J = 6.7 Hz, 2H), 7.09 (s,

1H), 7.15 (t, *J* = 7.3 Hz, 1H), 7.22 (t, *J* = 7.6 Hz, 1H), 7.38 (d, *J* = 8.5 Hz, 1H), 7.54 (d, *J* = 7.9 Hz, 1H), 8.04 (br s, 1H).

A Schlenk tube was charged with 8 (122 mg, 0.60 mmol), methyltriflate (148 mg, 0.90 mmol), and DCE (3 mL). The mixture was stirred for 2 h at 80 °C under N₂. After completion, the mixture was cooled to room temperature, quenched with saturated NaHCO₃, and extracted with EtOAc (10 mL \times 3). The organic layer was washed with saturated NaCl and dried over anhydrous MgSO₄. Evaporation of the solvent, followed by purification on silica gel (petroleum ether/ ethyl acetate), provided the product 9 as a white solid.

1-(*Methylthio*)-4,9-dihydro-3*H*-pyrido[3,4-b]indole (9). Isolated yield 117 mg (90%); white solid; mp 136–138 °C; R_f 0.36 (petroleum ether/ethyl acetate = 4:1 v/v). ¹H NMR (CDCl₃, 500 MHz): δ 2.53 (s, 3H), 2.90 (t, *J* = 8.2 Hz, 2H), 3.95 (t, *J* = 8.0 Hz, 2H), 7.16 (t, *J* = 7.3 Hz, 1H), 7.29 (t, *J* = 7.6 Hz, 1H), 7.39 (d, *J* = 8.5 Hz, 1H), 7.60 (d, *J* = 8.0 Hz, 1H), 8.29 (br s, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ 11.4, 19.7, 49.8, 112.0, 116.2, 120.1, 120.5, 124.7, 125.5, 128, 136.4, 156.2; HRMS (ESI-TOF, [M + H]⁺): calcd for C₁₂H₁₂N₂S, 217.0799, found 217.0792.

A round-bottom flask was charged with a mixture of anthranilic acid (69 mg, 0.5 mmol), 9 (108 mg, 0.5 mmol), and acetic acid (5 mL). The mixture was refluxed under stirring for 12 h. After cooling and evaporation of the solvent, the crude product was recrystallized from EtOH to furnish the rutaecarpine as a white solid.

*Rutaecarpine.*³⁰ Isolated yield 115 mg (80%); white solid; mp 259–261 °C (lit. 259–260 °C); R_f 0.56 (petroleum ether/ethyl acetate = 2:1 v/v). ¹H NMR (DMSO- d_6 , 500 MHz): δ 3.18 (t, J = 6.8 Hz, 2H), 4.45 (t, J = 6.8 Hz, 2H), 7.09 (t, J = 7.4 Hz, 1H), 7.26 (t, J = 7.6 Hz, 1H), 7.46–7.50 (m, 2H), 7.64 (d, J = 8.0 Hz, 1H), 7.68 (d, J = 8.1 Hz, 1H), 7.81 (t, J = 7.0 Hz, 1H), 8.16 (d, J = 8.0 Hz, 1H), 11.88 (s, 1H).

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.6b02605.

¹H and ¹³C NMR spectra of all new compounds (PDF) X-ray data for **3a** (CIF)

AUTHOR INFORMATION

Corresponding Authors

*E-mail: liming928@qust.edu.cn (M.L.). *E-mail: nick8110@163.com (W.-S.G.).

ORCID [©]

Ming Li: 0000-0003-4906-936X

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

This work was financially supported by the National Natural Science Foundation of China (21372137 and 21572110), the Natural Science Foundation of Shandong Province (ZR2014BM006), the Science and Technology Development Project in University of Shandong Province (J16LC12), the Applied Basic Research Project (Youth Special) of Qingdao (16-5-1-98-jch), and the Research Fund of State Key Laboratory for Marine Corrosion and Protection of Luoyang Ship Material Research Institute (No. KF160404).

REFERENCES

(1) (a) Costa, E. V.; Pinheiro, M. L. B.; Maia, B. H. L.; Soares, M. B.
 P.; Marques, N. S. F. A.; Ruiz, A. L. T. G.; Marchetti, G. M.; Carvalho,
 D. J. E.; Costa, C. O. S.; Galvao, A. F. C.; Lopes, N. P.; Koolen, H. H.
 F.; Bezerra, D. P.; Barison, A. J. Nat. Prod. 2016, 79, 1524–1531.
 (b) Zhang, Z.-H.; Zhang, H.-J.; Deng, A.-J.; Wang, B.; Li, Z.-H.; Liu,

Y.; Wu, L.-Q.; Wang, W.-J.; Qin, H.-L. J. Med. Chem. 2015, 58, 7557–7571. (c) Debono, A.; Capuano, B.; Scammells, P. J. J. Med. Chem. 2015, 58, 5699–5727. (d) Bentley, K. W. Nat. Prod. Rep. 2006, 23, 444–463.

(2) (a) Takeda, H.; Ishikawa, K.; Wakana, D.; Fukuda, M.; Sato, F.; Hosoe, T. J. Nat. Prod. 2015, 78, 2880–2886. (b) Bhadra, K.; Kumar, G. S. Med. Res. Rev. 2011, 31, 821–862.

(3) (a) Walker, K. A.; Boots, M. R.; Stubbins, J. F.; Rogers, M. E.; Davis, C. W. J. Med. Chem. **1983**, 26, 174–181. (b) Reddy, G. C. Tetrahedron Lett. **1995**, 36, 1001–1002.

(4) (a) Chiba, H.; Oishi, S.; Fujii, N.; Ohno, H. Angew. Chem., Int. Ed. 2012, 51, 9169–9172. (b) Allan, K. M.; Stoltz, B. M. J. Am. Chem. Soc. 2008, 130, 17270–17271.

(5) Bischler, A.; Napieralski, B. Ber. Dtsch. Chem. Ges. 1893, 26, 1903–1908.

(6) Pictet, A.; Spengler, T. Ber. Dtsch. Chem. Ges. 1911, 44, 2030–2036.

(7) (a) He, R.; Huang, Z.-T.; Zheng, Q.-Y.; Wang, C. *Tetrahedron Lett.* **2014**, *55*, 5705–5713. (b) He, L.; Nie, H.; Qiu, G.; Gao, Y.; Wu, J. Org. Biomol. Chem. **2014**, *12*, 9045–9053.

(8) (a) Zhu, Z.; Tang, X.; Li, X.; Wu, W.; Deng, G.; Jiang, H. J. Org. Chem. 2016, 81, 1401–1409. (b) Grigg, R.; Elboray, E. E.; Akkarasamiyo, S.; Chuanopparat, N.; Dondas, H. A.; Abbas-Temirek, H. H.; Fishwick, C. W. G.; Aly, M. F.; Kongkathip, B.; Kongkathip, N. Chem. Commun. 2016, 52, 164–166. (c) Li, J.; He, Y.; Luo, S.; Lei, J.; Wang, J.; Xie, Z.; Zhu, Q. J. Org. Chem. 2015, 80, 2223–2230.

(9) (a) Yoshida, Y.; Kurahashi, T.; Matsubara, S. Chem. Lett. 2011, 40, 1140–1142. (b) Iwayama, T.; Sato, Y. Chem. Commun. 2009, 5245–5247. (c) Korivi, R. P.; Cheng, C.-H. Org. Lett. 2005, 7, 5179–5182.

(10) (a) Zheng, D.; Li, S.; Wu, J. Org. Lett. 2012, 14, 2655–2657.
(b) Niu, Y.-N.; Yan, Z.-Y.; Gao, G.-L.; Wang, H.-L.; Shu, X.-Z.; Ji, K.-G.; Liang, Y.-M. J. Org. Chem. 2009, 74, 2893–2896. (c) Ding, Q.; Wu, J. Org. Lett. 2007, 9, 4959–4962.

(11) Yu, X.; Wu, J. J. Comb. Chem. 2009, 11, 895-899.

(12) (a) Zhao, D.; Lied, F.; Glorius, F. Chem. Sci. 2014, 5, 2869–2873. (b) Jayakumar, J.; Parthasarathy, K.; Cheng, C.-H. Angew. Chem., Int. Ed. 2012, 51, 197–200. (c) Guimond, N.; Fagnou, K. J. Am. Chem. Soc. 2009, 131, 12050–12051.

(13) (a) Villuendas, P.; Urriolabeitia, E. P. J. Org. Chem. 2013, 78, 5254–5263. (b) Chinnagolla, R. K.; Pimparkar, S.; Jeganmohan, M. Org. Lett. 2012, 14, 3032–3035. (c) Kornhaaβ, C.; Li, J.; Ackermann, L. J. Org. Chem. 2012, 77, 9190–9198.

(14) Ilardi, E. A.; Vitaku, E.; Njardarson, J. T. J. Med. Chem. 2014, 57, 2832–2842.

(15) Aradi, K.; Tóth, B. L.; Tolnai, G. L.; Novák, Z. Synlett 2016, 27, 1456–1485.

(16) (a) Hopkinson, M. N.; Sahoo, B.; Glorius, F. Adv. Synth. Catal.
2014, 356, 2794–2800. (b) Phipps, R. J.; McMurray, L.; Ritter, S.; Duong, H. A.; Gaunt, M. J. J. Am. Chem. Soc. 2012, 134, 10773–10776. (c) Vaddula, B. R.; Saha, A.; Leazer, J.; Varma, R. S. Green Chem. 2012, 14, 2133–2136.

(17) (a) Sinai, Á.; Vangel, D.; Gáti, T.; Bombicz, P.; Novák, Z. Org. Lett. 2015, 17, 4136–4139. (b) Walkinshaw, A. J.; Xu, W.; Suero, M. G.; Gaunt, M. J. J. Am. Chem. Soc. 2013, 135, 12532–12535. (c) Xu, Z.-F.; Cai, C.-X.; Liu, J.-T. Org. Lett. 2013, 15, 2096–2099. (d) Suarez, L. L.; Greaney, M. F. Chem. Commun. 2011, 47, 7992–7994.

(18) (a) Aradi, K.; Bombicz, P.; Novák, Z. J. Org. Chem. 2016, 81, 920–931. (b) Pang, X.; Chen, C.; Su, X.; Li, M.; Wen, L. Org. Lett. 2014, 16, 6228–6231. (c) Su, X.; Chen, C.; Wang, Y.; Chen, J.; Lou, Z.; Li, M. Chem. Commun. 2013, 49, 6752–6754. (d) Wang, Y.; Chen, C.; Peng, J.; Li, M. Angew. Chem., Int. Ed. 2013, 52, 5323–5327.

(19) Mukerjee, A.; Ashare, R. Chem. Rev. 1991, 91, 1-24.

(20) (a) He, Y.; Li, J.; Luo, S.; Huang, J.; Zhu, Q. Chem. Commun. 2016, 52, 8444–8447. (b) Tang, X.; Zhu, Z.; Qi, C.; Wu, W.; Jiang, H. Org. Lett. 2016, 18, 180–183.

(21) (a) Zhao, P.; Liu, Y.; Xi, C. Org. Lett. 2015, 17, 4388-4391.
(b) Zhao, P.; Yan, X.; Yin, H.; Xi, C. Org. Lett. 2014, 16, 1120-1123.

(22) (a) Gao, L.-P.; Ding, M.-W.; Sun, Y. Synth. Commun. 2006, 36, 1185–1191. (b) Gittos, M. W.; Robinson, M. R.; Verge, J. P.; Davies, R. V.; Iddon, B.; Suschitzky, H. J. Chem. Soc., Perkin Trans. 1 1976, 33–38.

(23) (a) Guo, W. S.; Li, S. L.; Tang, L.; Li, M.; Wen, L. R.; Chen, C. Org. Lett. **2015**, *17*, 1232–1235. (b) Wen, L. R.; Shen, Q. Y.; Guo, W. S.; Li, M. Org. Chem. Front. **2016**, *3*, 870–874.

(24) (a) Bowman, W. R.; Elsegood, M. R. J.; Stein, T.; Weaver, G. W. Org. Biomol. Chem. 2007, 5, 103–113. (b) Hamid, A.; Elomri, A.; Daich, A. Tetrahedron Lett. 2006, 47, 1777–1781.

(25) (a) Bielawski, M.; Aili, D.; Olofsson, B. J. Org. Chem. 2008, 73, 4602–4607. (b) Bielawski, M.; Zhu, M.; Olofsson, B. Adv. Synth. Catal. 2007, 349, 2610–2618.

(26) Liu, P.; Li, C.; Zhang, J.; Xu, X. Synth. Commun. 2013, 43, 3342–3351.

(27) Davies, R. V.; Iddon, B.; Paterson, T.; Pickering, M. W.; Suschitzky, H.; Gittos, M. W. J. Chem. Soc., Perkin Trans. 1 1976, 138–141.

(28) Raja, E. K.; Lill, S. O. N.; Klumpp, D. A. Chem. Commun. 2012, 48, 8141–8143.

(29) Park, S.; Hayes, B. L.; Marankan, F.; Mulhearn, D. C.; Wanna, L.; Mesecar, A. D.; Santarsiero, B. D.; Johnson, M. E.; Venton, D. L. J. Med. Chem. 2003, 46, 936–953.

(30) Huang, G.; Roos, D.; Stadtmuller, P.; Decker, M. Tetrahedron Lett. 2014, 55, 3607–3609.